

## New eco-efficient process to produce high-purity lithium hydroxide

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Lithium is a light alkali metal that is essential to produce lithium batteries and hence a key raw material in the process of the global green energy transition. Sources of lithium include lithium ores in pegmatitic and sedimentary rocks, brines in salt flats, oil or geothermal fields, and even seawater<sup>1)</sup>. Although this metal is present in about 60 parts per million (or grams per ton) in the earth's crust and is therefore more abundant than for instance lead, boron or tin<sup>2)</sup>, there are not too many lithium deposits worldwide with sufficiently high concentrations and quantities that allow a profitable exploitation.

At present, the extraction of lithium from salt brines is the most competitive practice in the market, due to its relatively low investment and operating costs. Most of the



Salar de Uyuni in Bolivia; Foto from Lucca Galuccia

important lithium deposits in salt flats worldwide are found in brines in the region known as the "lithium triangle", located in the territories of Argentina (salt flats of the Jujuy and Salta puna, and northern Catamarca), Bolivia (Uyuni and Coipasa salt flats) and Chile (Atacama and Maricunga salt flats)<sup>3)</sup>. In South America, Peru also has

significant reserves, present for example in the brines of the Bayovar and Sechura depressions in the Piura region<sup>4)</sup> or in the Falchani deposit of the Macusani project in the Puno region<sup>5)</sup>.

The growing demand for newer cathode technologies requires high-purity lithium hydroxide with low residual concentrations of ions such as sodium, calcium, magnesium or chloride. However, lithium from brines usually is obtained in the form of chloride, carbonate or sulfate, and there is demand for cost-effective and environmentally sound technologies for the production of battery grade lithium hydroxide.

At present there are few processes for the conversion of lithium salts into lithium hydroxide with the required quality, applying methods such as interionic exchange, selective extractions or electrolytic or electrodialytic procedures<sup>6-11)</sup>. Most of these known processes are characterized by low yields or in the case of electrochemical procedures by robustness deficiencies and frequently require harmful reagents and high amounts of water and energy. Particularly the high freshwater consumption required

for mining activities is causing socio-environmental conflicts due to its scarcity in the desert areas of the salt flats<sup>12)</sup>.

In response to these shortcomings and problems, K-UTEC Salt Technologies has developed an innovative process to produce lithium hydroxide in a quality suitable for battery production, avoiding the use of environmentally harmful chemicals and minimizing the use of freshwater<sup>13)</sup>. Particularly the water efficiency and robustness of the technology is considered an important advance compared to other lithium hydroxide production processes currently applied.

The new process consists of subjecting the pre-concentrated lithium solution to a sophisticated process using evaporation and cooling operations to eliminate most of the impurities present in order to crystallize high-purity lithium chloride. This product then is converted into lithium hydroxide using an innovative proprietary electro dialysis process. The process yield reaches



*Pilot Plant of K-UTEC Salt Technologies in Sondershausen*

almost 100% and commercially valuable by-products such as magnesium salts are also retained for commercialization. With 40% lower investment costs, a 30% reduction in operating costs and the commercialization of by-products, the technology is highly competitive and environmentally friendly<sup>13)</sup>. K-UTEC Salt Technologies has already optimized the process in its in-house pilot plant and process robustness has been proven during a six-month test run.

The importance of this innovation received official recognition through the silver medal for outstanding innovations in the chemical and polymer category in the framework of the IQ Innovation Award Central Germany 2020 (IQ Innovationspreis Mitteldeutschland 2020)<sup>14)</sup>, which confirms the importance and imminent practical application in the current context of global lithium production in the framework of the forthcoming global energy transition.

## Bibliography

- <sup>1)</sup> Oswald Eppers (2021), Salmueras de plantas de desalinización („Desalination plant reject brines“), *Horizonte Minero* N° 144 (<https://bit.ly/3xtpQBB>).
- <sup>2)</sup> David R. Lide (editor): CRC Handbook of Chemistry and Physics. 85. Edition, CRC Press, Boca Raton, Florida, 2005. Section 14, Geophysics, Astronomy, and Acoustics; Abundance of Elements in the Earth's Crust and in the Sea.
- <sup>3)</sup> Ministerio de Minería de Chile (2013), Comisión del Cobre, Mercado Internacional del Litio (<https://bit.ly/3edlvaE>).

- <sup>4)</sup> Thomas Schicht, Benjamin Wieser and Anne Allendor-Schicht (2013), K-UTEC Salt Technologies, Case study of a geophysical investigation with seismic refraction tomography and the OhmMapper to estimate the brine content of a Salar/Salmuera (<https://bit.ly/2CiGpYR>).
- <sup>5)</sup> a) Plateau Energy Metals (2016), Plateau Uranium Files Technical Report For Lithium and Potassium Resource Estimate and Potential By-product Value <https://bit.ly/2ZQw6mP>; b) American Lithium, American Lithium Reaches Agreement to Acquire Plateau Energy Metals and Consolidate Development-Stage Lithium Assets (<https://bit.ly/3vkiD4P>).
- <sup>6)</sup> Ryabtsev, A.D., Nemkov, N.M., Kotsupalo, N.P. et al. Preparation of High-Purity Lithium Hydroxide Monohydrate from Technical-Grade Lithium Carbonate by Membrane Electrolysis. Russian Journal of Applied Chemistry 77, 1108–1116 (2004).
- <sup>7)</sup> Kwang-Joo Kim (2008), Recovery of Lithium Hydroxide from Spent Lithium Carbonate using Crystallizations, Separation Science and Technology 43:2, 420-430.
- <sup>8)</sup> Li Li, Jing Ge, Renjie Chen, Feng Wu, Shi Chen, Xiaoxiao Zhang (2010), Environmental friendly leaching reagent for cobalt and lithium recovery from spent lithium-ion batteries, Waste Management 30:12, 2615-2621.
- <sup>9)</sup> N. P. Kotsupalo, V. P. Isupov, and A. D. Ryabtseva (2009), Prospects for the Use of Lithium-Bearing Minerals, Theor Found Chem Eng. 43:5, 800-809.
- <sup>10)</sup> Soyeong Joo, Hyun-Woo Shim, Jin-Ju Choi, Chan-Gi Lee, Dae-Guen Kim (2020), A Method of Synthesizing Lithium Hydroxide Nanoparticles Using Lithium Sulfate from Spent Batteries by 2-Step Precipitation Method, Korean Journal of Metals and Materials 58(4): 286-291.
- <sup>11)</sup> Guy Bourassa, Gary Pearse, Stephen Charles Mackie, Mykolas Gladkovas, Peter Symons, J David Genders y Jean-François Magnan, Nemaska Lithium Inc., Processes for preparing lithium hydroxide, European Patent EP2841623A4.
- <sup>12)</sup> Amanda Romero, José Aylwin y Marcel Didier (2019), Observatorio Ciudadano, Globalización de las empresas de energía renovable: Extracción de litio y derechos de los pueblos indígenas en Argentina, Bolivia y Chile (“Triángulo del Litio”) (<https://bit.ly/2BTBUeE>).
- <sup>13)</sup> a) Bernd Schultheis y Christoph Ney (2020), K-UTEC Salt Technologies, Energie der Erde Lithiumhydroxid für Batterien mit neuem Verfahren umweltschonend gewinnen, Gera IHK IQ Broschüre (<https://bit.ly/3fcs7aM>); b) Bernd Schultheis, resultados no publicados.
- <sup>14)</sup> Bundesforum Mittelstand (2020), Thüringer Innovation gewinnt beim 16. IQ Innovationspreis Mitteldeutschland (<https://bit.ly/2BVaRlr>).